

You have Downloaded, yet Another Great Resource to assist you with your Studies ©

Thank You for Supporting SA Exam Papers

Your Leading Past Year Exam Paper Resource Portal

Visit us @ www.saexampapers.co.za





## NATIONAL SENIOR CERTIFICATE

**GRADE 12** 

#### **PHYSICAL SCIENCES**

**COMMON TEST** 

**APRIL 2021** 

**MARKS** : 100

TIME : 2 Hours

This question paper consists of 11 pages and 2 data sheets

Copyright reserved Please turn over

#### **INSTRUCTIONS AND INFORMATION**

- 1. Write your centre number and examination number in the appropriate spaces on the ANSWER BOOK.
- This question paper consists of SEVEN questions. Answer ALL the questions in the ANSWER BOOK.
- 3. Start EACH question on a NEW page in the ANSWER BOOK.
- 4. Number the answers correctly according to the numbering system used in this question paper.
- 5. Leave ONE line between two sub-questions, for example between QUESTION 2.1 and QUESTION 2.2.
- 6. You may use a non-programmable calculator.
- 7. You may use appropriate mathematical instruments.
- 8. You are advised to use the attached DATA SHEETS.
- 9. Show ALL formulae and substitutions in ALL calculations.
- 10. Round off your final numerical answers to a minimum of TWO decimal places.
- 11. Give brief motivations, discussions et cetera where required.
- 12. Write neatly and legibly.

Copyright reserved Please turn over

#### QUESTION 1: MULTIPLE-CHOICE QUESTIONS

Various options are provided as possible answers to the following questions. Choose the answer and write only the letter (A-D) next to the question number (1.1-1.6) in the ANSWER BOOK, for example 1.11 D.

1.1 A car of mass 1000 kg pulls a trailer of mass 500 kg as shown.



According to Newton's third law of motion, which ONE of the following statements is TRUE?

- A The car and trailer pull each other with a force that is equal in magnitude but opposite in direction. Therefore, the net force is zero and the trailer cannot move.
- B The force that the car exerts on the trailer is greater than the force that the trailer exerts on the car. Therefore, the trailer moves forward.
- C The action force from the car is quicker than the reaction force from the trailer, so they move forward.
- D The action-reaction forces are equal in magnitude, but the force between the ground and wheels pushes them forward.
- 1.2 When the momentum of an object of constant mass is doubled, then its kinetic energy will be ...
  - A halved.
  - B doubled.
  - -C three times greater-
  - D four times greater.

(2)

(2)

- 1.3 Two objects are released from the same height at the same time. One object has TWICE the weight of the other. Neglecting friction, which ONE of the following statements is CORRECT for the motion?
  - A The heavier object hits the ground first
  - B The lighter object hits the ground first.
  - C Both objects hit the ground at the same time
  - On hitting the ground, the heavier object has a greater velocity than the lighter object

- 1.4 Which ONE of the following compounds is a KETONE?
  - A CH<sub>3</sub>COCH<sub>2</sub>CH<sub>3</sub>
  - B CH<sub>3</sub>COOCH<sub>2</sub>CH<sub>3</sub>
  - C CH<sub>3</sub>CH<sub>2</sub>CH<sub>2</sub>CHO
  - D CH<sub>3</sub>CH(OH)CH<sub>2</sub>CH<sub>3</sub> (2)
- 1.5 **P** and **Q** represent two organic compounds in the reactions below:

REACTION I: ethene HBr

**REACTION II**: P NaOH(aq) → Q heat

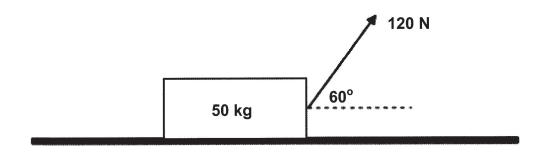
Which ONE of the following represents compound Q?

- A CH<sub>2</sub>CH<sub>2</sub>
- B CH<sub>3</sub>CH<sub>3</sub>
- C CH<sub>3</sub>CH<sub>2</sub>OH
- D  $CH_3CH_2Br$  (2)
- 1.6  $C_nH_{2n}O_2$  is the general formula for both . . .
  - A A ketone and an aldehyde.
  - B An ester and an aldehyde.
  - C A ketone and a carboxylic acid.
  - D An ester and a carboxylic acid. (2)

[12]

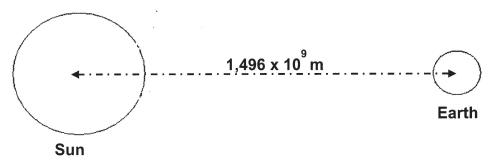
#### **QUESTION 2**

2.1 A box of mass of 50 kg, placed on a ROUGH horizontal surface, is pulled by a force of 120 N that acts at an angle of 60° to the horizontal. The box moves at a CONSTANT VELOCITY along the surface.



(2)2.1.1 State Newton's First Law of Motion in words.

- 2.1.2 Draw a free-body diagram of ALL the forces acting on the box while it (4) moves.
- 2.1.3 The box is travelling at constant velocity. What can be deduced about the (1) forces acting on the box?
- (3)2.1.4 Calculate the normal force acting on the box.
- 2.1.5 The angle of the applied force is now decreased. How will this affect the magnitude of the normal force? Choose from INCREASES, DECREASES or REMAINS THE SAME EXPLAIN the answer. (3)
- 2.2 The Sun and the Earth exert a gravitational force on each other.



2.2.1 State Newton's law of Universal Gravitation in words.

2.2.2 The mass of the Sun is 332 600 times greater than that of the Earth. The distance between the centres of the Sun and the Earth is 1,496 x 10 m. Calculate the gravitational force that the Sun exerts on the Earth.

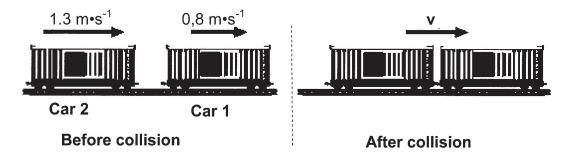
(4) [19]

(2)

(2)

#### **QUESTION 3**

A goods train is being assembled in a yard. Car 2, of mass of  $92 \times 10^3$  kg and moving with a velocity of  $1.3 \text{ m} \cdot \text{s}^{-1}$  to the right collides with Car 1 of mass  $65 \times 10^3$  kg that is moving to the right with a velocity of  $0.80 \text{ m} \cdot \text{s}^{-1}$ . The two cars are joined together after the collision. Ignore the effects of friction.

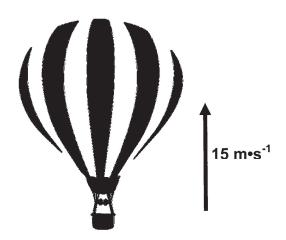


- 3.1 State the principle of conservation of momentum in words
- 3.2 Calculate the magnitude of the velocity of the cars after the collision. (4)
- 3.3 Determine, by means of appropriate calculations, whether the collision between the two cars is elastic or inelastic. (5)

  [11]

#### **QUESTION 4**

A hot air balloon is rising vertically at a constant velocity of 15 m·s<sup>-1</sup>. When the balloon is at some unknown height above the ground, a stone is released from the balloon. The stone is observed to hit the ground with a velocity of 45 m·s<sup>-1</sup>.



- 4.1 Define free fall (2)
- 4.2 Write down the magnitude and direction of the initial velocity of the stone when it is released (2)

- 4.3 Determine how high above the ground the hot air balloon was when the stone was released from it.
  - (3)
- 4.4 Determine how long the stone took to hit the ground after being released.
- (3)
- 4.5 Sketch the position-time graph for the entire motion of the stone from the moment it was released from the hot air balloon to the moment it hits the ground. Use the ground as a zero reference point.

Indicate the following on your graph:

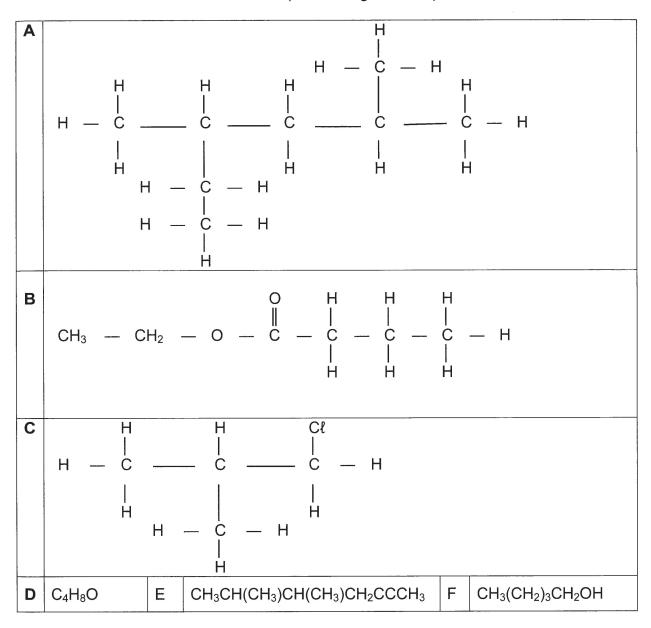
- Height when the stone is released.
- Time taken to hit the ground.

(4)

[14]

#### QUESTION 5 (Start on a new page.)

The letters A to G in the table below represent organic compounds.



Use the above table to answer the following questions:

5.1 Write down the:

5.1.3 Name of the homologous series to which compound B belongs. (1)

Copyright reserved Please turn over

5.2	Compo	ound E is a hydrocarbon.	
	5.2.1	Define the term hydrocarbon.	(2)
	5.2.2	Is compound E a saturated or unsaturated compound? Give a reason for the answer.	(2)
	5.2.3	Write down the general formula of the homologous series to which compound E belongs.	(1)
5.3	Write o	lown the IUPAC name of a POSITIONAL ISOMER of compound F.	(2)
5.4	D is the	e molecular formula of TWO functional isomers.	
	5.4.1	Define the term functional isomer.	(2)
	5.4.2	Draw the STRUCTURAL FORMULA of ONE of the FUNCTIONAL isomers of D.	(2)
	5.4.3	Write down the IUPAC name of the OTHER FUNCTIONAL isomer of D. (Hint: This is NOT the same compound as mentioned in Question 5.4.2)	(2) <b>[18]</b>

#### QUESTION 6 (Start on a new page.)

The boiling points of TWO organic compounds, A and B, were determined.

	FORMULA	MOLECULAR MASS (g.mol <sup>-1</sup> )
Α	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>2</sub> COOH	88,1
В	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>3</sub> CH <sub>2</sub> OH	88,1

6.1 State the definition of *boiling point*.

(2)

The following boiling points were obtained:

137 °C	163 °C
--------	--------

6.2 Write down the boiling point that is most likely to be that of compound A.

(2)

6.3 Explain FULLY how you arrived at the answer.

(4)

6.4 Write down the IUPAC name of the compound with the LOWER vapour pressure.

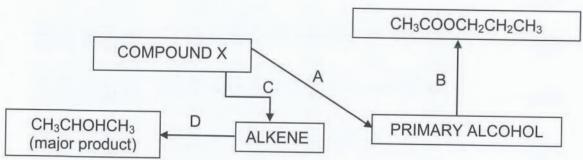
(2)

6.5 The boiling point of another compound, CH<sub>3</sub>(CH<sub>2</sub>)<sub>2</sub>CH<sub>2</sub>OH was also determined. How will the boiling point of this compound compare to that of compound B? Choose from GREATER THAN, LESS THAN or EQUAL TO. Explain the answer.

(3) **[13]** 

#### QUESTION 7 (Start on a new page.)

The flow diagram shows how COMPOUND X can be used to prepare other organic compounds. The letters A, B, C and D represent different organic reactions. Compound X is a HALOALKANE.



L	(major	product) ALKENE	
Use 7.1	the info Write 7.1.1	rmation in the flow diagram to answer the following questions: down the type of: Substitution reaction represented by A.	(1)
	7.1.2	Elimination reaction represented by C.	(1)
7.2	Write 7.2.1	down the type of reaction represented by: B.	(1)
	7.2.2	D.	(1)
7.3	Consid 7.3.1	der REACTION C. Write down TWO reaction conditions for this reaction.	(2)
	7.3.2	IUPAC name of the alkene formed.	(1)
7.4	Write	down the name or formula of the inorganic reactant for REACTION A.	(1)
7.5	React	ion B involves the reaction of an organic compound with a PRIMARY	
	7.5.1	Define the term <i>primary alcohol</i> .	(2)
	7.5.2	Write down the name or formula of the catalyst used in REACTION B.	(1)
	7.5.3	Write down the structural formula of the ORGANIC COMPOUND that reacts with the primary alcohol in REACTION B.	(2) [13]

[100]

TOTAL MARKS:

NSC-Grade 12

#### TABLE 3: THE PERIODIC TABLE OF ELEMENTS TABEL 3: DIE PERIODIEKE TABEL VAN ELE

At S 6 7  KEY/SLEUTEL  Electronegativity  Electronegativity  Electronegativity  Electronegativity  Electronegativity  Electronegativity  Electronegativity  Electronegativity  Benaderde re-  Benader re-  Benad	At 5 6 7  KEY/SLEUTEL  Benaderde re  Benader	10 11 12 13 14 15 16 17			mgeral He	Symbol 5 6 7 8 9	Simbool 2,0 B 2,5 C 3,0 N 3,5 O A,0	11 12 14 16 19	13 14 15 16 17	1,5 Ae 2,1 P 2,5 S 3,0 Ce	27 28 31 32 35,5	27 28 29 30 31 32 33 34 35	19 Co 18 Ni 19 Cu 16 Zn 16 Ga 18 Ge 20 As 24 Se 28 Br	59 59 63,5 65 70 73 75 79 80	45 46 47 48 49 50 51 52 53	22 Rh 22 Pd 1.9 Ag 1.7 Cd 1.7 In 1.8 Sn 1.9 Sb 21 Te 2.5 I	103 106 108 112 115 119 122 128 127	77 78 79 80 81 82 83 84 85	Ir Pt Au Hg 18 Te 18 Pb 19 Bi 2 Po 25 At	192 195 197 201 204 207		62 63 64 65 66 67 68 69 70	Sm Fil Gd Tb Dv Ho Fr Tm Yb	150 152 157	94 95 96 97 98 99 100 101 102	Pu Am Cm Bk Cf Es Fm Md	
School	Atomic number		<b>E</b>			S.	Ω	7		Ae	27	31	9,r <b>Ga</b>			<u>د</u>		2	<b>1</b> 6	204		99	2	163	86	ょ	
Feetronegativity   Elektronegativity   Elektronegativite    Elektroneg	(II)  KEY/SLEUTEL Atomic number  Atomic number  Atomic number  Approximate relative atomic mass  Benaderde relative atomic mass  Benaderde relative atomic mass  Benaderde relative atomic mass  Approximate relative atomic mass  Benaderde relative atomic mass  Benaderde relative atomic mass  Approximate relative atomic mass  Benaderde relative atomic mass  Benaderde relative atomic mass  Approximate relative atomic mass  Approximate relative atomic mass  Benaderde relative atomic mass  Approximate relative atomic mass  Benaderde relative atomic mass  Benaderde relative atomic mass  Approximate relative atomic mass  Benaderde relative atomic												1'9			۲'۱						65	T	159	-		 _
Atomic number  KEY/SLEUTEL  Atomic number  Atomic n	(II)  KEY/SLEUTEL Atomic number  Atomic number Atomic number  Atom	7											4'6			6°L						-					 _
3 4 5 6 7  KEY/SLEUTEL  Approximate  Electronegativity  Elektronegativity  Elektronegativity  Elektronegativity  Elektronegativity  Elektronegativity  Benaderde re  Benad	(ii)  KEY/SLEUTEL  Atol  KEY/SLEUTEL  Approximate re  Benaderde rela  24  25  Ca = Sc = T  Approximate re  Benaderde rela  26  27  Ca = Sc = T  Approximate re  Benaderde rela  27  Approximate re  Benaderde rela  28  39  40  41  42  43  55  55  55  55  88  89  91  92  73  74  75  88  89  89  91  72  73  74  75  88  89  89  91  72  73  74  74  75  88  89  89  89  91  74  75  88  89  89  89  76  77  78  78  78  79  79  79  79  79  70  70  70  70  71  71  71  71  71  71	9		<u>.</u>		vmbol	Simbool			nic mass	mmassa		8'₺			2,2						-					
3 4 5 6 7  KEY/SLEUTEL Atomorphish Electronegativity  Electronegativit	(ii)  KEY/SLEUTEL  Atol  KEY/SLEUTEL  Approximate re  Benaderde rela  24  20  21  22  Ca = Sc = T  Approximate re  Benaderde rela  24  40  45  40  41  42  43  88  89  40  41  42  43  55  55  55  88  89  91  92  92  90  91  92  Th  Pa  U  Th  Pa  U	6	,	c numb	тдета			<b>+</b>		ive aton	we atoo		8'L			2,2						-			-		 -
3 4 5 KEY/SLEL  KEY/SLEL  KEY/SLEL  KEY/SLEL  SC 1.5 12 2 2 3 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4	(ii)  KEY/(8  Be 9 12 Ng 24 20 Ca \(\frac{1}{12}\) Sr \(\frac{1}{12}\) Sr \(\frac{1}{12}\) Sr \(\frac{1}{12}\) Sr \(\frac{1}{12}\) Ba \(\frac{1}{139}\) Ra \(\frac{1}{12}\) Ra \(\frac{1}{139}\) Ra \(\frac{1}{12}\) Ra \(\frac{1}{139}\) Ra \(\frac{1}{12}\)			Atomi	A70C		6'L						8,1			2,2						-			-		
SC 1.5 TE FEE FEE TO SC 1.5 TO SC 1.	(ii)  KEY/S  (iii)  KEY/S  KEY/S  Be  9  12  Ca 7 2 1 22  Ca 7 8 7 7 1 6 7  88 89 91 9  56 57 72 7  Ba La 9 Hf 1  137 139 179 179  Ra Ac  Co C			111	7	ativity	atiwiteit			proxim	enaderd		3,1			6'١						-			-		 -
SC S	(ii) 3 4 4 (iii) 4 4 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8		)	VICI EIIT	ISECT	lectrone	ektronea			Ā	B	23	>	51	41		92	73	<u>ط</u>	181		-			-		
38 × 38 × 38 × 38 × 38 × 38 × 38 × 38 ×	(ii)  (iii)  4  Be 9 12 20 24 24 24 24 25 38 39 38 39 56 57 Ba 139 Ra Ac 226	4		X	Ž	ш	Ī					22	F	48	40		91			179		L		••••			 ]
	(E) 22	က	,									21	သင	45		>	83			139	83	Ac					

# DATA FOR PHYSICAL SCIENCES GRADE 12 PAPER 1 (PHYSICS) GEGEWENS VIR FISIESE WETENSKAPPE GRAAD 12 VRAESTEL 1 (FISIKA)

TABLE 1: PHYSICAL CONSTANTS / TABEL 1: FISIESE KONSTANTES

NAME / NAAM	SYMBOL / SIMBOOL	VALUE / WAARDE
Acceleration due to gravity Swaartekragversnelling	g	9,8 m·s <sup>-2</sup>
Universal gravitational constant Universele gravitasiekonstante	G	$6,67 \times 10^{-11} \text{ N} \cdot \text{m}^2 \cdot \text{kg}^{-2}$
Speed of light in a vacuum Spoed van lig in 'n vakuum	С	3,0 x 10 <sup>8</sup> m·s <sup>-1</sup>
Planck's constant Planck se konstante	h	6,63 x 10 <sup>-34</sup> J·s
Coulomb's constant Coulomb se konstante	k	9,0 x 10 <sup>9</sup> N·m <sup>2</sup> ·C <sup>-2</sup>
Charge on electron Lading op electron	e <sup>-</sup>	-1,6 x 10 <sup>-19</sup> C
Electron mass Elektronmassa	m <sub>e</sub>	9,11 x 10 <sup>-31</sup> kg
Mass of Earth Massa van Aarde	М	$5,98 \times 10^{24} \text{ kg}$
Radius of Earth Radius van Aarde	R <sub>E</sub>	6,38 × 10 <sup>6</sup> m

### TABLE 2: FORMULAE / TABEL 2: FORMULES MOTION / BEWEGING

$v_f = v_i + a \Delta t$	$\Delta x = v_i \Delta t + \frac{1}{2} a \Delta t^2 \text{ or/of } \Delta y = v_i \Delta t + \frac{1}{2} a \Delta t^2$
$v_f^2 = v_i^2 + 2a\Delta x \text{ or/of } v_f^2 = v_i^2 + 2a\Delta y$	$\Delta x = \left(\frac{v_i + v_f}{2}\right) \Delta t \text{ or/of } \Delta y = \left(\frac{v_i + v_f}{2}\right) \Delta t$

#### FORCE / KRAG

$F_{net} = ma$	p=mv	
$f_{s(max)} = \mu_s N$	$f_k = \mu_k N$	
$F_{net}\Delta t = \Delta p$ $\Delta p = mv_f - mv_i$	w=mg	
$F = \frac{Gm_1m_2}{r^2}$	$g = \frac{Gm}{r^2}$	

#### WORK, ENERGY AND POWER / ARBEID, ENERGIE EN DRYWING

$W = F\Delta x \cos \theta$	U= mgh or/of	$E_p = mgh$	
$K = \frac{1}{2} mv^2 \text{ or/of } E_k = \frac{1}{2} mv^2$	$W_{net} = \Delta K$	or/of	$W_{net} = \Delta E_k$
2 2	$\Delta K = K_f - K_i$	or/of	$\Delta E_k = E_{kf} - E_{ki}$
$W_{nc} = \Delta K + \Delta U$ or/of $W_{nc} = \Delta E_k + \Delta E_p$	$P = \frac{W}{\Delta t}$		
$P_{av} = F \cdot v_{av} / P_{gem} = F \cdot v_{gem}$			