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Controlled Test (Term 1) - 2024



Gauteng Department of Education Johannesburg North District (D10)

GRADE 12

PHYSICAL SCIENCES
CONTROLLED TEST 1
15 MARCH 2024

MARKING GUIDELINES

MARKS: 100

TIME: 2 Hours



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2

SECTION A: PHYSICS

QUESTION 1

1.1 C ✓✓ (2)

1.2 B ✓✓ (2)

1.3 D ✓✓
[6]

QUESTION 2

2.1 When a <u>resultant/net force</u> acts on an object, the object will accelerate in the direction of the force at an <u>acceleration directly proportional to the force</u> and <u>inversely proportional to the mass</u> of the object.

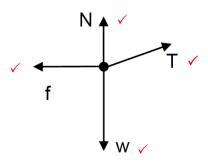
OR

The <u>resultant/net force</u> acting on an object is <u>equal to the rate of change</u> <u>of momentum</u> of the object (in the direction of the resultant/net force.) ✓✓

(2)

(4)

2.2



Notes

- Mark is awarded for label and arrow.
- Do not penalise for length of arrows.
- Deduct 1 mark for any additional force.
- If all forces are correctly drawn and labelled, but no arrows, deduct 1 mark.
- If force(s) do not make contact with body max ¾
- ACCEPT: If T is not shown but T|| and T₁ are shown, give 1 mark for both (However explain to learners that components will NO LONGER be accepted on free-body diagrams)



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(1)

3

2.3.1 ANY OF: ✓

The blocks/system are accelerating.

OR

The acceleration is changing / not zero / a \neq 0 (m·s⁻²) / a = 1,32 m.s⁻²

OR

Velocity is /increasing/changing/not constant

OR

F_{net} is not equal to zero (1)

2.3.2 $F_{net} = ma$ mg -T= ma $(2)(9,8) - T = 2(1,32) \checkmark$

> T = 16,96 N(3)

2.3.3 **POSITIVE MARKING FROM 2.3.2**

> F_{net} = ma $T_x - f_k = ma$ $T\cos 15^{\circ} - f_k = ma$

 $(16,96)\cos 15^{\circ} \checkmark - f_k = (8)(1,32) \checkmark$

(4) $f_k = 5.82 \text{ N}$ to the left \checkmark

2.4 Any ONE ✓

- Normal force changes/decreases
- The angle (between string and horizontal) changes/increases.
- The vertical component of the tension changes/increases

2.5 Yes ✓

> The <u>frictional force (coefficient of friction)</u>/ μ_k depends on the nature of the surfaces in contact. ✓

> > OR

The µ_k changes ✓ (2)[17]



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QUESTION 3

3.1 The total linear momentum of an isolated system remains constant ✓✓ (is conserved) in both magnitude and direction. (2 or zero)

Accept closed system however explain to learners a closed system will no longer be accepted (2)

3.2 **OPTION 1**

SOUTH AS POSITIVE

No. of passengers = $3 \cdot 120 - 2000 = 1 \cdot 120 \div 70 = 16$ passengers Therefore 16 - 12 = 4 passengers in excess \checkmark

OR

$$3\ 120 - 2000 - 12(70) = 280 \div 70 = 4$$
 passengers in excess (5)

OPTION 2

SOUTH AS POSITIVE

Let number of passengers = x

$$\begin{array}{c} \Sigma p_i = \Sigma p_f \\ m_1 v_{1i} + m_2 v_{2i} = (m_1 + m_2) v_f \\ \end{array}$$
 1 mark for any
$$(2000 + 70x)(+20) + (1\ 248)(-15) \checkmark = [(2000 + 70x) + 1\ 248](10) \checkmark \\ 40\ 000 + 1400x - 18\ 720 = 32\ 480 + 700x \\ 700x = 11\ 200 \\ x = 16 \checkmark$$
 Accept solution with North as Positive

Therefore 16 - 12 = 4 passengers in excess \checkmark

3.3 **POSITIVE MARKING FROM 3.2**

Mass of taxi = 3 120 kg

$$E_{ki} = \frac{1}{2} \text{mv}^2$$

$$E_{ki} = \frac{1}{2} \text{m}_1 \text{v}_{1i}^2 + \frac{1}{2} \text{m}_2 \text{v}_{2i}^2$$

$$= \frac{1}{2} (3 \ 120)(20)^2 + \frac{1}{2} (1 \ 248)(-15)^2 \checkmark$$

$$= 624 \ 000 + 140 \ 400$$

$$= 764 \ 400 \ J$$



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 $E_{kf} = \frac{1}{2}(m_1 + m_2)v_f^2$ $= \frac{1}{2}(3 \ 120 + 1 \ 248)(10)^2 \checkmark$ $= \frac{1}{2}(4 \ 368)(10)^2$ $= 218 \ 400 \ J$

 $E_{ki} \neq E_{kf}$ Therefore the collision is inelastic

(4)

- 3.4 During a collision, the crumple zone/ airbag:
 - The impulse/change in momentum/Δp remains constant
 - The contact time/Δt increases
 - The net force/force/F_{net} decreases

(3) **[14]**



6

QUESTION 4

4.1 9,8 m.s⁻² downwards ✓ (No marks if -9,8 or no direction included) (1)

4.2.1

OPTION 1 $v_f^2 = v_i^2 + 2a\Delta y$ ✓ $v_f^2 = (0)^2 + 2(-9,8)(-19,6)$ ✓ $v_f = -19,6 \text{ m} \cdot \text{s}^{-1}$ $v_f = 19,6 \text{ m} \cdot \text{s}^{-1}$ downwards ✓

Accept if downwards taken as positive

OPTION 2

$$\Delta y = v_i \Delta t + \frac{1}{2} a \Delta t^2$$

-19,6 = (0)² + $\frac{1}{2} (-9,8) \Delta t^2$
 $\Delta t = 2$

$$v_f = v_i + a\Delta t \checkmark$$

= 0 + (-9,8)(2) \checkmark
 $v_f = -19,6 \text{ m} \cdot \text{s}^{-1}$
 $\underline{v_f} = 19,6 \text{ m} \cdot \text{s}^{-1} \text{ downwards} \checkmark$

4.2.2 **OPTION 1**

For A:

$$\Delta x_A = v_i \Delta t + \frac{1}{2} a \Delta t^2$$

-19,6= 0 + $\frac{1}{2}(-9,8) \Delta t^2$
 $\Delta t = 2s$

$$\Delta t_A = \Delta t_B$$
 $\Delta x_B = v_i \Delta t + \frac{1}{2} a \Delta t^2 \checkmark$
 $-29.6 \checkmark = v_i(2) + \frac{1}{2} (-9.8)(2)^2 \checkmark$
 $v_i = -5 \text{ m} \cdot \text{s}^{-1}$
 $v_i = \frac{5 \text{ m} \cdot \text{s}^{-1} \text{ downwards}}{2} \checkmark$

OPTION 2

For A:

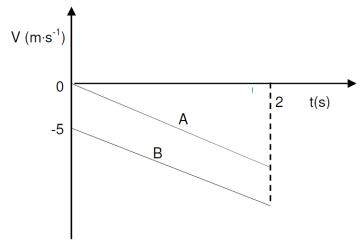
$$\frac{\nabla}{\nabla_{fA} = \nabla_{i}} + a\Delta t$$

$$-19,6 = 0 + (-9,8)\Delta t \checkmark$$

$$\Delta t = 2s$$

$$\Delta t_A = \Delta t_B$$
 $\Delta x_B = v_i \Delta t + \frac{1}{2} a \Delta t^2 \checkmark$
 $-29.6 \checkmark = v_i(2) + \frac{1}{2} (-9.8)(2)^2 \checkmark$
 $v_i = -5 \text{ m} \cdot \text{s}^{-1}$
 $v_i = \frac{5 \text{ m} \cdot \text{s}^{-1} \text{ downwards}}{2} \checkmark$

4.3 **POSITIVE MARKING FROM 4.2.2**



Marking criteria	Marks
A starts at 0 m·s ⁻¹ with negative gradient	✓
Graphs for A and B stop at 2s	√
B starts at -5 m·s ⁻¹ with negative gradient	✓
Lines parallel to each other	√



(3)

(5)

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SECTION B: CHEMISTRY

QUESTION 1

1.1 D ✓✓ (2)

1.2 C ✓✓ (2)

1.3 C ✓✓ (2) [6]

QUESTION 2

2.1 Unsaturated ✓

Any ONE: ✓

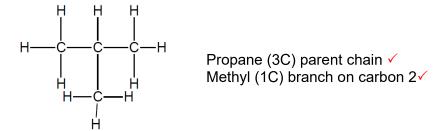
- Compound C has a <u>triple/multiple bond between C atoms</u>
- Compound C <u>does NOT contain the maximum number of H atoms bonded to C atoms</u>.
- Compound C <u>is an alkyne</u> (2)

2.3.1

Marking criteria:

- Whole structure correct 2/2
- Only functional group correct Max: 1/2
- If condensed or semi-structural formula used Max: 1/2 (2)
- 2.3.2 Hydroxyl (group) ✓ (Do NOT accept Alcohol) (1)
- 2.3.3 Butan-2-ol ✓ (Do NOT accept Butanol / Butan-1-ol) (1)

2.3.4



(2)

2.4.1 Haloalkane / Alkyl halide <

(1)



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(4)

8

2.4.2 3,5-√dibromo√octane √

Marking criteria:

- Octane (1 mark)
- Dibromo (1 mark)
- Substituents (dibromo) correctly numbered, hyphens, commas correctly used. (1 mark) (3)
- 2.5 $2\underline{C_4H_{10}}\checkmark + 13O_2 \rightarrow 8\underline{CO_2} + 10\underline{H_2O}$ both products \checkmark correct balancing \checkmark (3) [19]

QUESTION 3

- 3.1 150 kPa√ (1)
- 3.2.1 The <u>temperature</u> ✓ at which the <u>vapour pressure</u> of a liquid is <u>equal</u> to the <u>atmospheric (external) pressure</u> ✓ (2)
- 3.2.2 55°C ✓ (1)
- $3.3.1 \quad Z \checkmark$
- 3.3.2 <u>Carboxylic acids</u> have, in addition to London forces and dipole-dipole forces, <u>two sites for hydrogen bonding</u> between molecules. ✓

<u>Alcohols</u> have, in addition to London forces and dipole-dipole forces, <u>one</u> <u>site for hydrogen bonding</u> between molecules. ✓

<u>Ketones</u> have, in addition to London forces, <u>dipole-dipole</u> forces between molecules. ✓

Intermolecular forces in carboxylic acids is the strongest.

Most energy needed to overcome intermolecular forces in carboxylic acid hence the lowest vapour pressure.

3.3.4 Propanone√ (Accept: propan-2-one) (1) [10]



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QUESTION 4

4.1 Secondary ✓ The C atom bonded to the <u>OH group</u> is bonded to TWO other C atoms ✓ (2) 4.2.1 Elimination of water /H₂O ✓ (1) 4.2.2 Sulphuric acid or Phosphoric acid ✓ 4.3.1 Hydrogenation√ (1) 4.3.2 Dehydrohalogenation / dehydrobromination ✓ (1) (1) 4.4.1 Substitution (hydrolysis)✓ 4.4.2 Dilute base / Dilute sodium hydroxide (NaOH) ✓ Moderate temperature/(mild) heat ✓ (2) (2) 4.4.3 2√-bromobutane ✓ 4.5 (3)Accept H₂ (instead of H-H) 4.6 butane√ (1) [15]

TOTAL SECTION B: [50 MARKS]
[TOTAL = 100 MARKS]



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COGNITIVE LEVELS FOR PHYSICAL SCIENCE GRADE 12 – CONTROLLED TEST 2024

		COGNITIVE LEVELS					
QUESTION		1	2	3	4		
	MARKS	Recall	Comprehension	Analysis Application	Evaluation Synthesis		
		(15 %)	(40 %)	(35 %)	(10 %)		
SECTION A: PHYSICS							
1.1	2		2				
1.2	2			2			
1.3	2			2			
2.1	2	2					
2.2	4	4	4				
2.3.1 2.3.2	3	1		3			
2.3.3	4			4			
2.4	1			1			
2.5	2		2	·			
3.1	2	2					
3.2	5			5			
3.3	4		4				
3.4	3			3			
4.1	1	1					
4.2.1	3		3		-		
4.2.2	5 4			4	5		
4.3	4	SECT	ION B: CHEMISTRY				
1.1	2	3LC1	ION B. CHEWISTKI	2			
1.2	2		2				
1.3	2				2		
2.1	2			2			
2.2.1	1		1				
2.2.2	1			1			
2.2.3	1		1				
2.2.4	1		1				
2.3.1	1		2 1				
2.3.3	1		1				
2.3.4	2		2				
2.4.1	1		1				
2.4.2	3				3		
2.5	3		3				
3.1	1		1				
3.2.1	2	2					
3.2.2	1		1				
3.3.1	1			1			
3.3.2 3.3.3	4 1			4	1		
4.1	2	2			ı		
4.2.1	1	1					
4.2.2	1	1					
4.3.1	1	1					
4.3.2	1			1			
4.4.1	1		1				
4.4.2	2		2				
4.4.3	2		2				
4.5	3		3				
4.6	1		1				
TOTAL	100	13	41	35	11		
%	100%	13%	41%	35%	11%		

