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PROVINCIAL ASSESSMENT

GRADE 12

PHYSICAL SCIENCES

CONTROL TEST

MARCH 2025

MARKS: 100

TIME: 2 hours

This paper consists of 16 pages including DATASHEET and
the PERIODIC TABLE

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INSTRUCTIONS AND INFORMATION

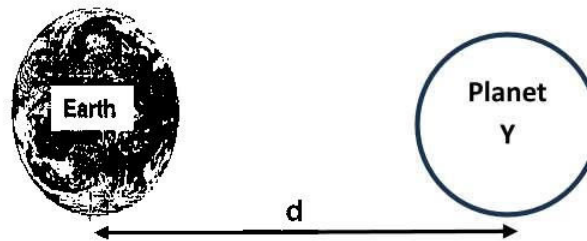
1. Write your name and other information in the appropriate spaces on the ANSWER BOOK.
2. This question paper consists of SEVEN questions. Answer ALL questions in the ANSWER BOOK.
3. Number the answers correctly according to the numbering system used in this question paper.
4. Leave ONE line between two subquestions, e.g. between QUESTION 2.1 and QUESTION 2.2
5. You may use a non-programmable pocket calculator.
6. Show ALL formulae and substitutions in ALL calculations.
7. Round off your FINAL numerical answers to a minimum of TWO decimal places where applicable.
8. Give brief motivationa, discussions,etc. where required.
9. You are advised to use the attached DATA SHEETS.
10. Write neatly and legibly.



QUESTION 1

Various options are provided as possible answers to the following questions. Each question has only ONE correct answer. Choose the answer and write only the letter (A–D) next to the question numbers (1.1 to 1.16) in the ANSWER BOOK, e.g. 1. 11 D.

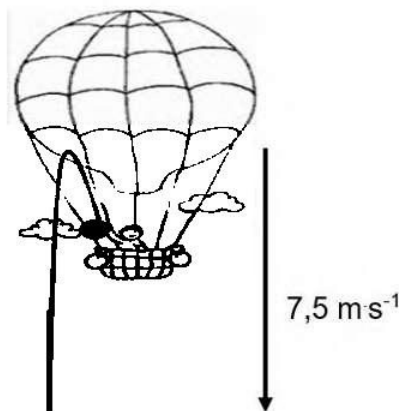
- 1.1 The gravitational force that the earth and Planet Y exerts on each other is F when there is a distance d between their centres. What would be the magnitude of the force experienced between the two when the distance between their centres is $3d$?



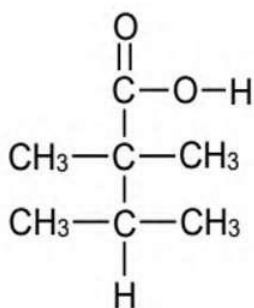
- A $9F$.
- B $\frac{1}{3} F$.
- C $\frac{1}{9} F$.
- D $3 F$. (2)
- 1.2 The change in momentum of an object can be equated to its...
- A change in velocity.
- B product of mass and velocity.
- C change in kinetic energy.
- D product of net force and change in time. (2)



- 1.3 A hot air balloon is descending at a velocity of $7,5 \text{ m}\cdot\text{s}^{-1}$ when a boy throws a plastic disc upwards at a velocity of $4 \text{ m}\cdot\text{s}^{-1}$. What is the initial velocity of plastic disc?



- A $11,5 \text{ m}\cdot\text{s}^{-1}$ downwards.
 B $3,5 \text{ m}\cdot\text{s}^{-1}$ downwards.
 C $4 \text{ m}\cdot\text{s}^{-1}$ downwards.
 D $7,5 \text{ m}\cdot\text{s}^{-1}$ downwards. (2)
- 1.4 The condensed structural formula of a compound is shown below.



Which one of the following is the correct IUPAC name for this compound?

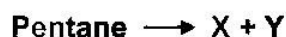
- A 2,2,1,1-tetramethylpropanoic acid
 B 2,2,3- trimethylbutanoic acid
 C 1,1,2,2-tetramethylpropanoic acid
 D 3,2,2-trimethylbutanoic acid (2)



1.5 Which one of the following *functional groups* has the strongest intermolecular forces.

- A Hydroxyl and Formyl
- B Formyl and Carbonyl
- C Hydroxyl and Carboxyl
- D Carboxyl and Carbonyl (2)

1.6 The following cracking reaction takes place in high temperatures and pressures.



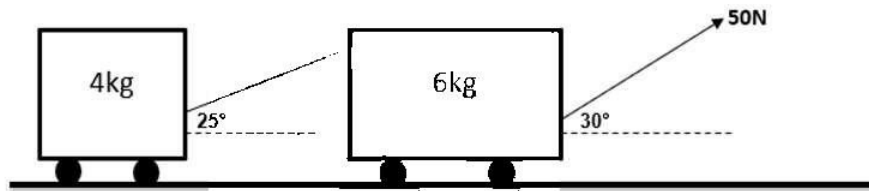
Which of the following compounds are likely to be the products of the above reaction?

- A Ethene + propane
 - B Ethane + propane
 - C Propene + ethene
 - D Ethanol + propanol (2)
- [12]



QUESTION 2

Two toy cars of mass 4 kg and 6 kg respectively are connected by a light inextensible rope inclined at 25° to the horizontal as shown. The 6 kg toy car is pulled by a 50 N applied force inclined at 30° to the horizontal. The kinetic frictional force experienced by the 4 kg and 6 kg toy cars are 10 N and 15 N respectively.



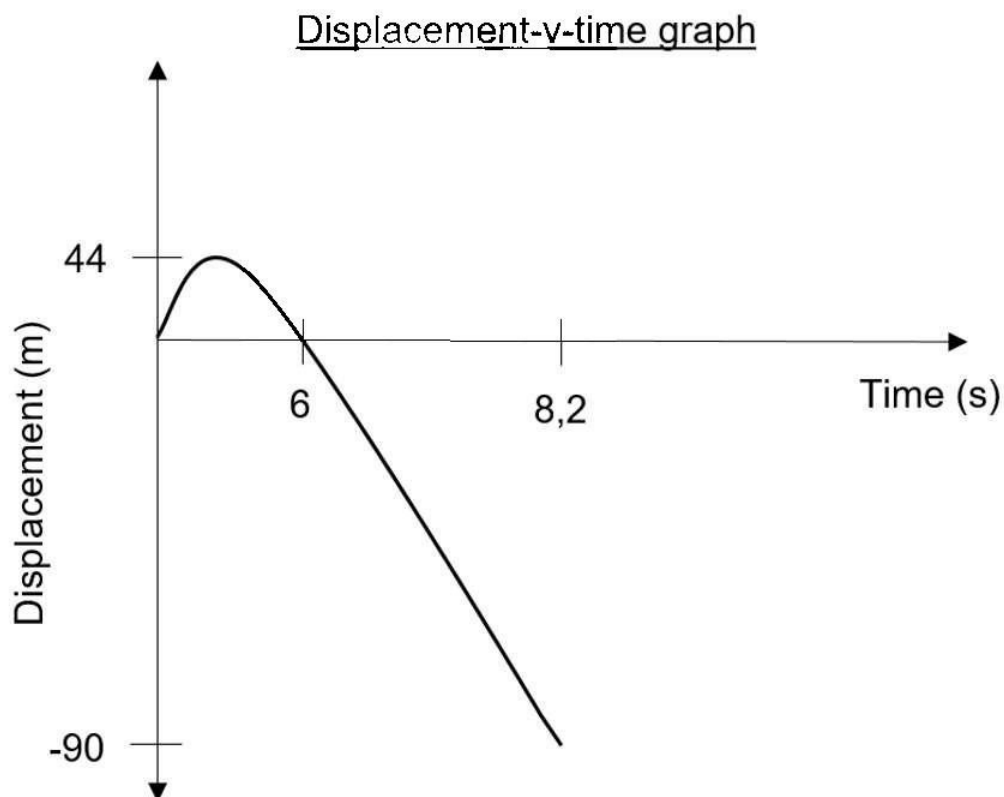
- 2.1 State *Newton's second law of motion* in words. (2)
 - 2.2 Draw a labelled free-body diagram showing all forces acting on the 6 kg toy car. (5)
 - 2.3 Calculate the acceleration experienced by the toy cars. (6)
 - 2.4 Calculate the tension on the rope. (2)
- [15]**



QUESTION 3

A girl was standing on top of a hotel building playing with a ball of mass 1,2 kg. She projected the ball upwards and failed to catch it on its way down. The ball fell to the ground.

The position versus time graph below shows the motion of the ball from the moment it was projected until it reached the ground.



- 3.1 Define the term *Free fall*. (2)
- 3.2
- 3.2.1 What is the height of the building from which the ball was thrown? (1)
- 3.2.2 What is the **acceleration** of the ball at maximum height? (1)
- 3.3 Using equations of **motion** only, Calculate the:
- 3.3.1 Initial **velocity** of the ball. (3)
- 3.3.2 Velocity **with** which the ball hits the ground. (3)



- 3.3.3 While the ball is in contact with the ground, it experiences a Net force of 130 N and then bounces off the ground with a velocity of $16,5 \text{ m.s}^{-1}$.
Calculate the time the ball was in contact with the ground. (4)
- 3.4 Draw a velocity versus time graph for the motion of the ball from the moment it was thrown until it strikes the ground.
On the graph show the following:
- Velocity at which the ball was thrown and
 - Velocity at which the ball hits the ground. (3)
- [17]**



QUESTION 4

A group of learners were investigating the *rate of change in momentum* using two trolleys, Trolley **A** with mass 2 kg and Trolley **B** with a mass of 5 kg were placed on a flat frictionless surface. Trolley **A** moves easterly towards a stationary trolley **B**, the two trolleys collide and move together after the collision.

The table below shows the initial velocities of trolley **A** and the collision time for two collision trails that took place between the trolleys.

Trolley A	Initial velocity ($\text{m}\cdot\text{s}^{-1}$)	Contact time(s)
Trail 1	1,5	0.4
Trail 2	2	0.25

- 4.1 State the term **represented** by the words in italics in the above statement. (1)
- 4.2 Calculate the **magnitude** of the velocity of the two trolleys after collision in **trail 1**. (4)
- 4.3 Calculate the net force experienced by trolley **B** in **trial 2**. (5)
- 4.4 Will the magnitude of the forces that the trolleys exert on each other in trial 1 be EQUAL to the magnitude of the forces in Trail 2? Choose from YES or NO. Give a reason for the answer. (2)
- [12]**



QUESTION 5

Consider organic compounds represented by letters **A** to **F** in the table below.

A	$ \begin{array}{ccccccc} & \text{H} & & \text{CH}_3 & & \text{H} & & \text{CH}_2\text{CH}_3 \\ & & & & & & & \\ \text{H} & - \text{C} & - & \text{C} & - & \text{C} & - & \text{C} & - \text{H} \\ & & & & & & & \\ & \text{H} & & \text{CH}_3 & & \text{H} & & \text{CH}_2\text{CH}_3 \end{array} $	B	Butanoic acid
C	$ \begin{array}{ccccccc} & \text{H} & & \text{H} & & \text{O} & & \text{H} \\ & & & & & & & \\ \text{H} & - \text{C} & - & \text{C} & - & \text{C} & - & \text{C} & - \text{H} \\ & & & & & & & \\ & \text{H} & & \text{H} & & & & \text{H} \end{array} $	D	$\text{CH}_3\text{CH}(\text{CH}_3)\text{CH}_2\text{CHClCH}_3$
E	$\text{CH}_3\text{CH}_2\text{CH}_2\text{COH}$	F	$ \begin{array}{ccccccc} & \text{H} & & & \text{O} & & \text{H} & & \text{H} \\ & & & & & & & & \\ \text{H} & - \text{C} & - & \text{O} & - & \text{C} & - & \text{C} & - & \text{C} & - \text{H} \\ & & & & & & & & \\ & \text{H} & & & & & \text{H} & & \text{H} \end{array} $

Use the information in the table to answer the questions that follow.

5.1 Write down a letter that represents the following.

5.1.1 A Haloalkane. (1)

5.1.2 A compound that has a Carboxyl group as a functional group. (1)

5.2 Write down the :

5.2.1 IUPAC name of compound **A**. (3)

5.2.2 General formula of compound **A**. (1)

5.2.3 IUPAC name of compound **E**. (2)

5.2.4 Structural formula of compound **D** . (3)

5.3 Compound **C** and **E** are isomers.

5.3.1 Define the term *structural isomer*. (2)

5.3.2 Identify the type of Isomerism between the two compounds? (1)



5.4 Compound **F** is an ester. Write down the...

5.4.1 IUPAC name of this ester. (2)

5.4.2 structural formula of the alcohol that was used to produce this ester. (2)

5.4.3 inorganic compound that is produced in the process of producing the ester. (1)
[19]



QUESTION 6

Learners use the following compounds to investigate the factors that affect the vapour pressure of organic compounds.

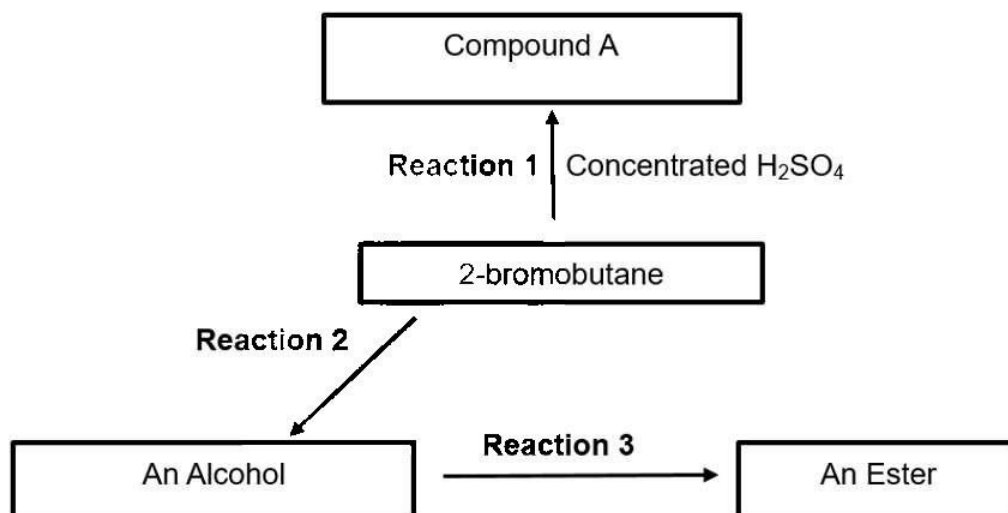
	COMPOUND	VAPOUR PRESSURE (kPa at 25°C)
A	2-methyl propane	150
B	Butane	83
C	Propan-1-ol	54
D	Ethanoic acid	26

- 6.1 Define the term *Vapour pressure*. (2)
- 6.2 Which compound **A** or **B** has a higher boiling point?
Explain your answer. (4)
- 6.3 Explain the difference in vapour pressures of compound **C** and **D** by referring to the TYPE OF INTERMOLECULAR FORCES present in each compound. (4)
- [10]**



QUESTION 7

In the flow diagram below 2-bromobutane undergoes different reactions as illustrated. Answer the questions that follow.



7.1 Consider **reaction 1**, write down the:

7.1.1 Type of reaction that takes place. (1)

7.1.2 IUPAC name of compound A which is a major product. (2)

7.2 Consider **reaction 2**, write down the:

7.2.1 Type of reaction that takes place. (1)

7.2.2 Structural formula of the alcohol. (2)

7.2.3 Reaction conditions (2)

7.3 Consider **reaction 3**

7.3.1 Write down the homologous series to which the OTHER organic reactant in this reaction belongs. (1)

7.3.2 The ester contains 6,6% of Hydrogen(H), 40% of Carbon(C) and 53,33% of Oxygen (O). The molar mass of the ester is $150\text{g}\cdot\text{mol}^{-1}$. Use a calculation to determine its molecular formula. (6)
[15]

TOTAL: 100



**DATA FOR PHYSICAL SCIENCES GRADE 12
PAPER 1 (PHYSICS)**

TABLE 1: PHYSICAL CONSTANTS

NAME	SYMBOL	VALUE
Acceleration due to gravity	g	$9,8 \text{ m}\cdot\text{s}^{-2}$
Universal gravitational constant	G	$6,67 \times 10^{-11} \text{ N}\cdot\text{m}^2\cdot\text{kg}^{-2}$
Speed of light in a vacuum	c	$3,0 \times 10^8 \text{ m}\cdot\text{s}^{-1}$
Planck's constant	h	$6,63 \times 10^{-34} \text{ J}\cdot\text{s}$
Coulomb's constant	k	$9,0 \times 10^9 \text{ N}\cdot\text{m}^2\cdot\text{C}^{-2}$
Charge on electron	e	$-1,6 \times 10^{-19} \text{ C}$
Electron mass	m_e	$9,11 \times 10^{-31} \text{ kg}$
Mass of the Earth	M	$5,98 \times 10^{24} \text{ kg}$
Radius of the Earth	R_E	$6,38 \times 10^6 \text{ m}$

TABLE 2: FORMULAE**MOTION**

$v_f = v_i + a \Delta t$	$\Delta x = v_i \Delta t + \frac{1}{2} a \Delta t^2$ OR $\Delta y = v_i \Delta t + \frac{1}{2} a \Delta t^2$
$v_f^2 = v_i^2 + 2a\Delta x$ OR $v_f^2 = v_i^2 + 2a\Delta y$	$\Delta x = \left(\frac{v_i + v_f}{2} \right) \Delta t$ OR $\Delta y = \left(\frac{v_i + v_f}{2} \right) \Delta t$

FORCE

$F_{\text{net}} = ma$	$p = mv$
$f_s^{\text{max}} = \mu_s N$	$f_k = \mu_k N$
$F_{\text{net}} \Delta t = \Delta p$ $\Delta p = mv_f - mv_i$	$w = mg$
$F = \frac{Gm_1 m_2}{d^2}$ OR $F = \frac{Gm_1 m_2}{r^2}$	$g = \frac{GM}{d^2}$ OR $g = \frac{GM}{r^2}$



CHEMISTRY

$n = \frac{m}{M}$	$n = \frac{N}{N_A}$
$c = \frac{n}{V}$ <i>or/of</i> $c = \frac{m}{MV}$	$n = \frac{V}{V_m}$



TABLE 3: PERIODIC TABLE

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
(I)	(II)	(III) (IV) (V) (VI) (VII) (VIII)															
1 1,0 H	3 1,5 Li	4 2,0 Be	9 4,0 F	10 1,9 Ne	20 3,0 N	14 2,5 O	16 3,5 S	17 3,0 Cl	18 35,5 Ar	36 2,8 Kr	54 2,5 Xe	86 2,5 Rn					
11 0,9 Na	12 2,4 Mg	23 3,9 Al	24 2,0 Si	27 1,5 P	31 2,4 S	32 2,5 Cl	33 2,0 Ar	35 2,8 Br	36 79 Kr	49 1,7 In	50 1,8 Sn	81 1,8 Tl	82 2,0 Pb	127 2,5 I	128 2,0 Po	173 2,5 Yb	175 103 Lu
19 0,8 K	20 0,6 Ca	39 0,8 Rb	40 0,6 Sr	51 0,8 Nb	52 0,8 Mo	55 0,8 Tc	56 0,8 Ru	59 0,8 Rh	60 0,8 Pd	79 0,8 Ag	80 0,8 Cd	101 0,8 Os	102 0,8 Ir	131 0,8 Xe	132 0,8 La	173 0,8 Yb	175 0,8 Lu
37 0,8 Rb	38 0,8 Sr	87 0,7 Fr	88 0,7 Ra	92 0,8 Nb	93 0,8 Mo	96 0,8 Tc	101 0,8 Ru	103 0,8 Rh	106 0,8 Pd	108 0,8 Ag	112 0,8 Cd	190 0,8 Os	192 0,8 Ir	131 0,8 Xe	132 0,8 La	173 0,8 Yb	175 0,8 Lu
55 0,7 Cs	56 0,7 Ba	87 0,7 Fr	88 0,7 Ra	73 0,8 Ta	74 0,8 W	75 0,8 Re	76 0,8 Os	77 0,8 Ir	78 0,8 Pt	79 0,8 Au	80 0,8 Hg	190 0,8 Os	192 0,8 Ir	131 0,8 Xe	132 0,8 La	173 0,8 Yb	175 0,8 Lu
87 0,7 Fr	88 0,7 Ra	133 0,7 Cs	137 0,7 Ba	181 0,8 Ta	184 0,8 W	186 0,8 Re	190 0,8 Os	192 0,8 Ir	195 0,8 Pt	197 0,8 Au	201 0,8 Hg	204 0,8 Tl	207 0,8 Pb	127 2,5 I	128 2,0 Po	173 2,5 Yb	175 103 Lu
103 0,8 Lr	104 0,8 Ac	173 0,8 Yb	175 0,8 Lu	101 0,8 Md	102 0,8 No	100 0,8 Fm	99 0,8 Es	98 0,8 Cf	97 0,8 Bk	96 0,8 Cm	95 0,8 Am	94 0,8 Pu	93 0,8 Np	92 0,8 U	91 0,8 Pa	90 0,8 Th	89 0,8 Ac

